

## COPPER-BASED ALLOY CASTING IN WHICH GRAINS ARE REFINED

### BACKGROUND OF THE INVENTION

#### 1. Technical Field

[0001] The present invention relates to a copper-based alloy casting in which grains are refined after melt-solidification, and particularly, to a Cu-Zn-Si alloy casting.

#### 2. Related Art

[0002] It is well known that, like a common metallic material, the proof stress of a copper-based alloy is improved by grain refinement, and it is told that the strength of the copper-based alloy is inversely proportional to the square root of the grain size on the basis of the Hall-Petch theory.

[0003] Basically, the grains of the copper-based alloy are refined as follows: (A) grains are being refined during the melt-solidification of the copper-based alloy, or (B) grains are refined by performing deformation process such as rolling or heat treatment on the melt-solidified copper alloy (ingot such as slurry or the like; casting such as diecast or the like; and hot forged parts or the like), in which stacking energy such as distortion energy or the like acts as a driving force.



[0004] Zr is a well-known element contributing to the grain refinement in both (A) and (B) cases.

[0005] However, in (A) case, the effect of Zr on the grain refinement during the melt-solidification is considerably influenced by the other components and the amount thereof, whereby a desired level of grain refinement cannot be achieved.

[0006] As a result, grains are, in general, refined like (B) case, that is, heat treatment is performed on a melt-solidified ingot, casting or the like, and then the alloy is distorted for grain refinement.

[0007] These are disclosed in JP-B-38-20467 and JP-A-2004-100041.

[0008] In JP-B-38-20467, solution heat treatment and 75% cold-working are performed on a copper alloy containing Zr, P and Ni in order to examine the mean grain size. The publication illustrates that the grain size decreases as the amount of Zr increases by showing that the mean grain sizes are 280  $\mu\text{m}$  (no Zr contained), 170  $\mu\text{m}$  (0.05% of Zr contained), 50  $\mu\text{m}$  (0.13% of Zr contained), 29  $\mu\text{m}$  (0.22% of Zr contained) and 6  $\mu\text{m}$  (0.89% of Zr contained) respectively. In addition, the publication suggests that the optimum amount of Zr is 0.05 to 0.3%, considering adverse effects induced when Zr is contained excessively.

[0009] JP-A-2004-233952 discloses that, in a copper alloy



containing 0.15 to 0.5% of Zr, grains can be reduced about 20  $\mu\text{m}$  or less in the mean grain size by performing solution heat treatment and deformation process, which is to add distortion to the alloy.

[0010] However, performing the above treatment and process after casting for the purpose of grain refinement, like (B) case, causes cost to increase, and sometimes it is impossible to perform deformation process for distortion-adding due to the shapes of casting parts.

[0011] Therefore, it is preferable that grains are being refined during the melt-solidification of the copper-based alloy like (A) case.

[0012] However, as described above, in (A) case, since the effect of Zr on the grain refinement during the melt-solidification is considerably influenced by the other elements and the amount thereof, the grain refinement is not necessarily promoted by the increase of Zr amount. In addition, the affinity of Zr to oxygen is so high that Zr is susceptible to oxidation when dissolved and added in the air, whereby the process yield is very low. As a result, a considerable amount of raw material must be injected during pouring even when a casting contains a small amount of Zr.

[0013] Meanwhile, if too many oxides are generated during dissolving, the oxides can enter the mold during pouring, and thus casting defects occur. In order to prevent the



oxides from being generated, it can be a good method to dissolve and cast the alloy in vacuum or inert gas atmosphere, however, it leads to cost rise.

[0014] Furthermore, Zr is an expensive element. Therefore it is preferable, from an economic viewpoint, to contain a small amount of Zr.

[0015] As a result, a copper-based alloy casting, the amount of Zr of which is reduced as much as possible, and the grains of which are refined after melt-solidification, is required.

[0016] In a Cu-Zn-Si alloy, Si improves the mechanical properties or the like. However, due to Si, cracks and cavities easily occur during the melt-solidification, whereby shrinkage cavities become large, and casting defects such as shrinkage cavity or the like easily occur. The above phenomena are mainly induced by the following facts: as the amount of Si increases, the solidification temperature range between liquidus temperature and solidus temperature is widened, and the thermal conductivity decreases. Furthermore, it can be found out from the solidified structure of Cu-Zn-Si alloy in the related art that dendrites are shaped, and the arms of the dendrites hinder the removal of porosities generated in the casting. Therefore, the shrinkage cavities remain in the casting, and large shrinkage cavities are generated locally.



[00017] The present inventor found out that, if the grains are being refined during the melt-solidification, shrinkage stress generated at the final stage of solidification decreases, and stresses exerting solid phases are dispersed. Therefore, cracks and cavities seldom occur, and the arms of the dendrites are cut. In addition, porosities are easily removed, and shrinkage cavities are smoothly generated. As a result, a casting without casting defect can be obtained.

#### SUMMARY

[0018] It is an object of an aspect of the invention to provide a copper-based alloy in which grains are being refined during the melt-solidification, and more specifically, to provide a Cu-Zn-Si alloy casting in which grains are refined to be 100  $\mu\text{m}$  or less in the mean grain size after melt-solidification.

[0019] In order to solve the above problems, a first copper-based alloy casting according to the invention includes 69 to 88% of Cu, 2 to 5% of Si, 0.0005 to 0.04% of Zr, 0.01 to 0.25% of P by mass, and a remainder includes Zn and inevitable impurities, and satisfies  $60 \leq \text{Cu} - 3.5 \times \text{Si} - 3 \times \text{P} \leq 71$ . Further, mean grain size after melt-solidification is 100  $\mu\text{m}$  or less, and  $\alpha$ ,  $\kappa$  and  $\gamma$ -phases occupy more than 80% of phase structure.

[0020] A second copper-based alloy casting according to



the invention further includes, in addition to the composition of the first copper-based alloy casting, at least one element selected from a group consisting of 0.001 to 0.2% of Mg, 0.003 to 0.1% of B, 0.0002 to 0.01% of C, 0.001 to 0.2% of Ti and 0.01 to 0.3% of rare earth element as a grain-refining element, and satisfies  $60 \leq \text{Cu} - 3.5 \times \text{Si} - 3 \times \text{P} - 0.5 \times [\text{i}] + 0.5 \times [\text{ii}] \leq 71$ . Here, [i] is a group consisting of Mg and B, and [ii] is a group consisting of C, Ti and rare earth element.

[0021] A third copper-based alloy casting according to the invention further includes, in addition to the composition of the first copper-based alloy casting, at least one element selected from a group consisting of 0.02 to 1.5% of Al, 0.2 to 4.0% of Mn and 0.01 to 0.2% of Cr as a strength and wear resistance-improving element, and satisfies  $60 \leq \text{Cu} - 3.5 \times \text{Si} - 3 \times \text{P} - 1.8 \times \text{Al} + a \times \text{Mn} + 0.5\text{Cr} \leq 71$  ( $a = 2$  in a case that Mn is contained more than 0.5% and satisfies  $0.2 \times \text{Si} \leq \text{Mn} \leq 2.0 \times \text{Si}$ , and  $a = 0.5$  in the other cases).

[0022] A fourth copper-based alloy casting according to the invention further includes, in addition to the composition of the second copper-based alloy casting, at least one element selected from a group consisting of 0.02 to 1.5% of Al, 0.2 to 4.0% of Mn and 0.01 to 0.2% of Cr as a strength and wear resistance-improving element, and satisfies  $60 \leq \text{Cu} - 3.5 \times \text{Si} - 3 \times \text{P} - 0.5 \times [\text{i}] + 0.5 \times [\text{ii}]$



-  $1.8 \times \text{Al} + a \times \text{Mn} + 0.5\text{Cr} \leq 71$  ( $a = 2$  in a case that Mn is contained more than 0.5% and satisfies  $0.2 \times \text{Si} \leq \text{Mn} \leq 2.0 \times \text{Si}$ , and  $a = 0.5$  in the other cases).

[0023] First to fourth copper-based alloy castings according to the invention can further include at least one element selected from a group consisting of 0.1 to 2.5% of Sn, 0.02 to 0.25% of Sb and 0.02 to 0.25% of As as a corrosion resistance-improving element, and at least one element selected from a group consisting of 0.004 to 0.45% of Pb, 0.004 to 0.45% of Bi, 0.03 to 0.45% of Se and 0.01 to 0.45% of Te as a machinability-improving element.

[0024] In this specification, 'mean grain size after melt-solidification' means the mean grain size measured after melt-solidification of the copper-based alloys of predetermined compositions, on which no deformation process such as rolling and heat treatment are performed.

#### BRIEF DESCRIPTION OF THE DRAWINGS

[0025] Fig. 1 is a photomicrograph ( $\times 350$ ) showing a phase structure of Specimen No. 9 of embodiments;

[0026] Fig. 2 is a photomicrograph ( $\times 350$ ) showing a phase structure of Specimen No. 103 of comparative examples;

[0027] Fig. 3 is macro photograph and photomicrograph ( $\times 75$ ) showing a metal structure of a cross section of Specimen No. 9 of the embodiments;



[0028] Fig. 4 is macro photograph and photomicrograph ( $\times 75$ ) showing a metal structure of a cross section of Specimen No. 10 of the embodiments;

[0029] Fig. 5 is macro photograph and photomicrograph ( $\times 75$ ) showing a metal structure of a cross section of Specimen No. 6 of the embodiments;

[0030] Fig. 6 is macro photograph and photomicrograph ( $\times 75$ ) of a metal structure showing a cross section of Specimen No. 112 of the comparative examples;

[0031] Fig. 7 is macro photograph and photomicrograph ( $\times 75$ ) showing a metal structure of a cross section of Specimen No. 110 of the comparative examples;

[0032] Fig. 8 is macro photograph and photomicrograph ( $\times 75$ ) showing a metal structure of a cross section of Specimen No. 103 of the comparative examples;

[0033] Fig. 9A is a graph showing a relationship between the amount of Zr and the mean grain size within the extent of  $64 \leq \text{Cu} - 3.5 \times \text{Si} - 3 \times \text{P} \leq 67$ ;

[0034] Fig. 9B is a graph of Fig. 9A, wherein the amount of Zr is expressed in logarithmic scale;

[0035] Fig. 10 is a photomicrograph ( $\times 75$ ) showing the shape of dendrite in Specimen No. 8 of the comparative example;

[0036] Fig. 11 is a photomicrograph ( $\times 75$ ) showing the shape of dendrite in Specimen No. 115 of the comparative



example;

[0037] Fig. 12 is a photomicrograph ( $\times 75$ ) showing the shape of dendrite in Specimen No. 110 of the comparative example;

[0038] Figs. 13A to 13C are views showing final solidification portions in Tatur Shrinkage Test. Fig. 13A is a view of final solidification portion evaluated 'good', Fig. 13C is a view of final solidification portion evaluated 'bad', and Fig. 13B is a view of final solidification portion evaluated 'fair';

[0039] Figs. 14A to 14C are photographs showing cross-sections of Specimen No. 9 of the embodiments. Fig. 14A is a photograph without magnification, Fig. 14B is a photograph ( $\times 3.5$ ), and Fig. 14C is a photograph ( $\times 18$ ); and

[0040] Figs. 15A to 15C are photographs showing cross-sections of Specimen No. 109 of the comparative example. Fig. 15A is a photograph without magnification, Fig. 15B is a photograph ( $\times 3.5$ ), and Fig. 15C is a photograph ( $\times 18$ ).

#### DESCRIPTION OF THE EMBODIMENTS

[0041] First of all, the reason why the amount of each of alloy components composed of copper-based alloy castings of the invention is defined and the relationships of each of the components will be described.

[0042] Hereinafter, '%' of the alloy components means



percent by mass.

Furthermore, the relationships are as follows;

Expression (1):  $Cu - 3.5 \times Si - 3 \times P$

Expression (2):  $Cu - 3.5 \times Si - 3 \times P - 0.5 \times [i] + 0.5 \times [ii]$

Expression (3):  $Cu - 3.5 \times Si - 3 \times P - 1.8 \times Al + a \times Mn + 0.5 \times Cr$

Expression (4):  $Cu - 3.5 \times Si - 3 \times P - 0.5 \times [i] + 0.5 \times [ii] - 1.8 \times Al + a \times Mn + 0.5 \times Cr$

[0043] Copper-based alloy castings according to the invention contain 69 to 88% of Cu, 2 to 5% of Si, 0.0005 to 0.04% of Zr and 0.01 to 0.25% of P, and the remainder is Zn and inevitable impurities.

Cu: 69 to 88%

[0044] Cu is the main component of the alloys. The grains are not necessarily refined in all copper-based alloy castings even when Zr and P are contained. The present inventor, as described below, found out that the grains can be refined considerably by adding a small amount of Zr when the amount of Si and P follow predetermined relationships.

[0045] In order to obtain various characteristics such as mechanical properties, corrosion resistance or the like as an industrial material, more than 69% of Cu is contained. However, if more than 88% of Cu is contained, the grain refinement is hindered, whereby the maximum amount of Cu is



88%. In addition, it is preferable to contain 70 to 84% of Cu, more preferable to contain 71 to 79.5% of Cu, and most preferable to contain 72 to 79% of Cu.

Si: 2 to 5%

[0046] Si decreases the stacking fault energy of the alloys and refines the grains considerably when contained with Zr, P, Cu, and Zn. In this case, 2% or more of Si must be contained. However, if more than 5% of Si is contained, the grain refinement saturates or tends to be hindered even when added with Cu and Zn. In addition, the ductility of the castings decreases. Also, the thermal conductivity decreases, and the solidification temperature range is widened, whereby the machinability deteriorates. Si also improves the fluidity of the molten alloy, prevents oxidation of the molten alloy, and decreases the melting point of the molten alloy. Furthermore, Si improves the corrosion resistance, particularly dezincification corrosion resistance and stress corrosion cracking resistance. Si also improves the machinability and the mechanical strength such as tensile strength, proof stress, impact strength, fatigue strength or the like. The above effects induce synergy effect for the grain refinement of the castings. It is preferable to contain 2.2 to 4.8% of Si, more preferable to contain 2.5 to 4.5% of Si, and most preferable to contain 2.7 to 3.7% of Si in order to induce the synergy effect.



Zr: 0.0005 to 0.04%

[0047] Zr is an important component for the grain refinement of the castings. As described below, if the amounts of Cu, Si and P follow predetermined relationships, the grains are effectively refined when 0.0005% or more of Zr is contained. The grains are refined more effectively at the amount of 0.0008% or more, most effectively at the amount of 0.0010% or more, and the grain refinement saturates at the amount of 0.0095%.

[0048] Meanwhile, the affinity of Zr to oxygen and sulfur is so strong that it is difficult to add Zr within a targeted narrow composition range. Therefore, considering that copper alloy castings are, in general, fabricated with recycled and scrapped materials in the air, a considerable amount of Zr must be added. Meanwhile, the inventor found out that, when 0.05% or more of Zr is contained in Cu-Zn-Si alloys, the grains are not being effectively refined during the melt-solidification. As a result, the maximum amount of Zr is defined at 0.04%. Zirconium oxide is easily formed in the casting and thus robust castings are difficult to obtain as the amount of Zr increases. In addition, since Zr is an expensive metal, it is not desired, from an economic viewpoint, to contain a large amount of Zr. Therefore, it is preferable to contain 0.0290% or less of Zr, more preferable to contain 0.0190%, and most preferable to



contain 0.0095%, at which the grain refinement saturates, as described above. That is, considering the effect of Zr on matters other than the grain refinement, the optimal amount of Zr is 0.0010 to 0.0095%.

P: 0.01 to 0.25%

[0049] P is, like Zr, an important component for the grain refinement of the castings. When contained with Zr, P refines the grains considerably. In addition, P increases the fluidity of the molten alloy, precipitates  $\kappa$ ,  $\gamma$  and  $\beta$ -phases, to be described below, more finely, and improves the corrosion resistance. P shows the above effects when contained 0.01%. However, when P is contained too much, a low-melting point intermetallic compound is formed, and thus the alloy becomes brittle. Therefore, the maximum amount of P is defined at 0.25%, considering easy fabrication of the castings. Meanwhile, even though varying with the mixing ratio with the mixing amount of Zr and the mixing amount or ratio of Cu, Zn, Si of the matrix, the amount of P is preferably in the range of 0.02 to 0.20%, more preferably in the range of 0.03 to 0.16%, and most preferably in the range of 0.04 to 0.12%.

Zn: Remainder

[0050] Zn is a main component composing the copper-based alloy castings of the invention with Cu and Si. Zn decreases the stacking fault energy of the alloy and refines



the grains of the castings. In addition, Zn induces the increase in the fluidity of the molten alloy, the decrease in the melting point, the prevention of Zr oxidation, the increase in corrosion resistance, and the increase in machinability. Furthermore, Zn improves the mechanical strength such as tensile strength, proof stress, impact strength, fatigue strength or the like. Therefore, Zn composes the alloy with the above components.

[0051] Meanwhile, Cu, Si and P among the components of the copper-based alloy castings of the invention are further required to satisfy the value of Expression (1):  $Cu - 3.5 \times Si - 3 \times P$  is in the range from 60 to 71 as well as the above-mentioned conditions.

[0052] The expression is derived experimentally from the measured sizes of melt-solidified grains and conditions refining the grains of the copper-based alloy castings about 100  $\mu m$  or less in the mean grain size. Even though Expression (1) will be described in detail later, it is preferable that Expression (1) have the value in the range of 62.5 to 68.5, and more preferable in the range of 64 to 67.

[0053] Furthermore, in the copper-based alloy castings of the invention, it is preferable that  $P/Zr$  be in the range of 0.8 to 250,  $Si/Zr$  be in the range of 80 to 6000, and  $Si/P$  be in the range of 12 to 220 in order to achieve the desired



level of grain refinement.

[0054] P/Zr is preferably in the range of 1.5 to 150, more preferably in the range of 2 to 100, and most preferably in the range of 4 to 50. Si/Zr is preferably in the range of 100 to 5000, more preferably in the range of 120 to 3500, and most preferably in the range of 300 to 1500. Si/P is preferably in the range of 16 to 160, more preferably in the range of 20 to 120, and most preferably in the range of 25 to 80.

[0055] Meanwhile, on the assumption that Cu, Zn, Si, P, P/Zr, Si/Zr, Si/P, Expression (1) and phase structure are as claimed, Zr, particularly when added with P, increases the stacking fault density of a melt-solidified material and causes crystal nuclei to be generated faster than grain growth, whereby micronization of a melt-solidified material, specifically, the grains of the castings can be realized.

[0056] A second copper-based alloy casting according to the invention can further contain at least one element selected from a group consisting of 0.001 to 0.2% of Mg, 0.003 to 0.1% of B, 0.0002 to 0.01% of C, 0.001 to 0.2% of Ti and 0.01 to 0.3% of rare earth element, in addition to the components of the first copper-based alloy casting, as a grain refinement-promoting element.

[0057] Mg considerably decreases the loss of Zr due to sulfur and oxygen, which are mixed from recycled or scrapped



materials such as inferior products, scrapped products, wire scrap, press scrap of copper alloy, chips, sprue runner generated during processing, sink head, lasher, mill ends or the like, and exists in the form of manganese sulfide and/or (manganese) oxide in the molten alloy so as to prevent the presence of Zr not contributing to the grain refinement, thereby contributing to the grain refinement. That is, if Mg is added before Zr, Mg becomes MgS and MgO, whereby Mg decreases the amount of S and O in the molten alloy and render Zr do its role effectively. Therefore, it is preferable to contain at least 0.001% of Mg in the alloy.

[0058] B, C, Ti and rare earth elements contribute to the grain refinement. Therefore, it is preferable to contain at least 0.003% of B, 0.0002% of C, 0.001% of Ti, and 0.01% of rare earth elements in order for the elements to effectively work. In this case, the rare earth elements (REM) mean fourteen kinds of lanthanoid elements, including La, Ce or the like as well as Sc and Y.

[0059] On the other hand, if Mg, B, C, Ti and rare earth elements are added too much, their effects saturate and the fluidity of the molten alloy deteriorates. Therefore, the maximum amounts of Mg, B, C, Ti and rare earth elements are defined at 0.2%, 0.1%, 0.01%, 0.2% and 0.3%, respectively.

[0060] In addition, the above elements are related with the effect of Zr on the grain refinement and influence



Expression (1) of the first copper-based alloy casting, whereby, considering each of the effects of Mg, B, C, Ti and rare earth element, the amounts of the respective elements are adjusted to satisfy that the values of Expression (2):  $Cu - 3.5 \times Si - 3 \times P - 0.5 \times [i] + 0.5 \times [ii]$  are in the range of 60 to 71. In this case, [i] is a group composed of Mg and B, and [ii] is a group composed of C, Ti, and rare earth element.

[0061] A third copper-based alloy casting according to the invention can further contain at least one element selected from a group consisting of 0.02 to 1.5% of Al, 0.2 to 4.0% of Mn and 0.01 to 0.2% of Cr, in addition to the components of the first copper-based alloy casting, in order to increase the strength and wear resistance of the first copper-based alloy casting.

[0062] When containing the above elements, the alloys in which grains are refined have excellent strength and wear resistance.

[0063] Al strengthens the matrix so as to improve the strength and wear resistance. Therefore, it is preferable to contain 0.02% or more of Al, and more preferable to contain 0.1% or more of Al. However, if Al is contained too much, the elongation deteriorates, whereby the maximum amount of Al is defined at 1.5%.

[0064] Mn is combined with Si to form a Mn-Si



intermetallic compound and contributes to improve the wear resistance. Therefore, it is preferable to contain 0.2% or more of Mn, and more preferable to contain 0.5% or more of Mn. However, if more than 4.0% of Mn is contained, the above effect saturates, and the fluidity of the molten alloy deteriorates, whereby Si, which is useful for the grain refinement, is consumed due to the formation of Mn-Si intermetallic compound. Therefore, the maximum amount of Mn is defined at 4.0%, and it is preferable to contain 3.5% or less of Mn.

[0065] Meanwhile, in order to suppress the consumption of Si, which is useful for the grain refinement, it is preferable that the amount of Si satisfy  $2.3 + 1/3\text{Mn} \leq \text{Si} \leq 3.5 + 1/3\text{Mn}$ , and more preferable that it satisfy  $2.6 + 1/3\text{Mn} \leq \text{Si} \leq 3.4 + 1/3\text{Mn}$ .

[0066] Cr is partially dissolved in the matrix and partially forms a fine intermetallic compound with Si so as to improve the wear resistance. Therefore, it is preferable to contain 0.01% or more of Cr. However, if Cr is contained too much, the Cr-Si compound is coarsened and the above effect saturates. Therefore, the maximum amount of Cr is defined at 0.2%.

[0067] In addition, Al, Mn and Cr are related with the effect of Zr on the grain refinement and influence Expression (1) of the first copper-based alloy casting.



Therefore, considering the effect of Al, Mn and Cr, the amounts of the respective elements are adjusted to satisfy that the values of Expression (3):  $Cu - 3.5 \times Si - 3 \times P - 1.8 \times Al + a \times Mn + 0.5Cr$  are in the range of 60 to 71. In this case,  $a = 2$  in a case that Mn is contained 0.5% or more and satisfies  $0.2 \times Si \leq Mn \leq 2.0 \times Si$ , and  $a = 0.5$  in the other cases.

[0068] A fourth copper-based alloy casting according to the invention can further contain at least one element selected from a group consisting of 0.02 to 1.5% of Al, 0.2 to 4.0% of Mn and 0.01 to 0.2% of Cr, in addition to the components of the second copper-based alloy casting, in order to increase the strength and wear resistance of the second copper-based alloy casting.

[0069] As described above, the elements are related with the effect of Zr on the grain refinement and influence Expression (2) of the second copper-based alloy casting. Therefore, considering the effects of Al, Mn and Cr, the amounts of the respective elements are adjusted to satisfy that the values of Expression (4):  $Cu - 3.5 \times Si - 3 \times P - 0.5 \times [i] + 0.5 \times [ii] - 1.8 \times Al + a \times Mn$  are in the range of 60 to 71. In this case,  $a = 2$  in a case that Mn is contained 0.5% or more and satisfies  $0.2 \times Si \leq Mn \leq 2.0 \times Si$ , and  $a = 0.5$  in the other cases.

[0070] First to fourth copper-based alloy castings can



further contain at least one element selected from a group consisting of 0.1 to 2.5% of Sn, 0.02 to 0.25% of Sb and 0.02 to 0.25% of As as a corrosion resistance-improving element.

[0071] Containing the above elements can increase the corrosion resistance of the alloys in which grains are refined.

[0072] Sn improves erosion · corrosion resistance, seawater resistance. In particular, the synergy effect of Sn with Si forms Si and Sn-rich protective coatings in a corrosive liquid so as to lead to an excellent corrosion resistance. Therefore, it is preferable to contain 0.1% or more of Sn. However, if more than 2.5% of Sn is contained, segregations easily occur, and casting cracks easily occur since Sn is a low-melting point metal. In addition, the ductility deteriorates. Therefore, the maximum amount of Sn is defined at 2.5%, and it is preferable to contain in the range of 0.2 to 0.9% of Sn.

[0073] Sb and As improve dezincification corrosion resistance. Therefore, it is preferable to contain 0.02% or more of Sb and/or As. However, if Sb and/or As are contained too much, segregations easily occur, and casting cracks easily occur since Sb and As are low-melting point metals. In addition, the ductility may deteriorate. Therefore, the maximum amounts of Sb and As are defined



0.25% respectively.

[0074] In addition, the copper-based alloy castings according to the invention can further contain at least one element selected from a group consisting of 0.004 to 0.45% of Pb, 0.004 to 0.45% of Bi, 0.03 to 0.45% of Se and 0.01 to 0.45% of Te as a machinability-improving element.

[0075] When containing the above elements, the alloys in which grains are refined have excellent machinability.

[0076] The machinability can be improved by containing 0.004% of Pb, 0.004% of Bi, 0.03% of Se and/or 0.01% of Te.

[0077] Meanwhile, since Pb, Bi, Se and Te have negative influences on the human body, and Bi, Se and Te are rare elements, the maximum amounts of Pb, Bi, Se and Te are defined at 0.45% respectively. When the casting of the invention is used for valves for potable water, metal fitting for water-supplying and drainage or the like, it is preferable that the maximum amount of the above element be 0.2% or less.

[0078] In the copper-based alloy casting of the invention, it is allowed to contain impurities that are inevitably contained during the melting of raw material of the alloy. However, if Fe and Ni, as impurities, are contained too much, they consume Zr and P, which are useful for the grain refinement, so as to hinder the grain refinement. Therefore, when Fe and/or Ni is contained, amount thereof is defined at



0.5% or less. In addition, it is preferable to contain 0.25% or less of Fe and Ni, and more preferable to contain 0.15% or less of Fe and 0.2% or less of Ni.

[0079] In phase structures of the copper-based alloy castings of the invention,  $\alpha$ ,  $\kappa$  and  $\gamma$ -phases are adjusted to occupy 80% or more of the structure, and it is preferable that the above three phases be adjusted to occupy 100% of the structure. Meanwhile, the concentrations of Si in  $\kappa$  and  $\gamma$ -phases are higher than that in  $\alpha$ -phase, and at least one of  $\beta$ ,  $\mu$  and  $\delta$ -phases occupies the remainder when the three phases do not occupy 100% of the structure.

[0080] In order to obtain a phase structure, 80% or more of which are occupied by  $\alpha$ ,  $\kappa$  and  $\gamma$ -phases, it is required to optimize the casting conditions such as pouring temperature, cooling rate or the like. This is also a condition required to reduce the mean size of the melt-solidified grains to be about 100  $\mu\text{m}$  or less.

[0081] In order to obtain an industrially satisfying machinability without addition of Pb, it is preferable that  $\kappa + \gamma$ -phase occupy 5 to 85% of the phase structure, and it is more preferable that the phase occupy 10 to 80% of the phase structure. On the other hand, when the other phases occupy more than 20% of the structure, a primary crystal is not  $\alpha$ -phase, whereby the grain refinement cannot be achieved. In addition, machinability, corrosion-resistance, elongation



and impact strength deteriorate. Particularly, in order to obtain excellent dezincification corrosion resistance, elongation and machinability, it is desirable that  $\beta$ -phase occupy 10% or less of the structure.

[0082] Hereinafter, the transformation of phase structure during the melt-solidification will be described in detail, since closely related with Expressions (1) to (4).

[0083] It is desirable that the primary crystal be  $\alpha$ -phase during solidification. That is, if the solid phase is  $\alpha$ -phase while the crystal nuclei are generated, the grain refinement is further expedited. In this case, the values of Expressions (1) to (4) are equivalent to 62.5. In addition, it is most preferable that the amount of  $\alpha$ -phase primary crystal be 20 to 30% or more, and, in this case, the values of Expressions (1) to (4) are equivalent to 64. Furthermore, in the actual solidification process, if peritectic or eutectic reaction occurs during solidification,  $\alpha$ -phase solid can exist in Cu-Zn-Si alloy when the solidification is complete, whereby the above fact becomes a condition for the grain refinement, and the values of Expressions (1) to (4) are equivalent to 62.5. Even though the mean grain size is large even at compositions having the values close but not equal to 62.5, the grains are refined. Then, Expressions (1) to (4) have values of 60 as the minimum.



[0084] Meanwhile, Expressions (1) to (4) have the values of 71 due to the following facts: the grain refinement becomes more difficult as the amount of added Zn decreases, peritectic reaction does not occur in a practical non-equilibrium state during solidification, and the machinability deteriorates. In addition, the solidification temperature range is widened. If the solidification temperature range is widened, solid-phase granular coalescences are easily generated, and thus dendrites have shapes similar to a tree. Furthermore, even though the grains are refined to a certain degree, cracks and cavities easily occur, and the number and size of shrinkage cavities and shrinkage cavities increase.

[0085] The grains are best refined when phases other than  $\alpha$ -phase, mainly  $\beta$ ,  $\kappa$  or  $\gamma$ -phase are crystallized or precipitated after solidification. That is, as the number of  $\alpha$ -phase primary crystals increases, the grains are coalesced with one another, whereby the primary crystals are shaped like dendrites having grown arms. If phases other than  $\alpha$ -phase such as  $\beta$ ,  $\gamma$  and  $\kappa$ -phase are crystallized or precipitated in order to prevent the above phenomenon, the growth of  $\alpha$ -phase grains during solidification and cooling from high temperature are suppressed, and thus micronization can be realized. For example, if eutectic reaction occurs during solidification, the micronization of grains can be



realized. In order for the second phase to exist during solidification, it is desirable that the values of Expressions (1) to (4) have 68.5 or less, most desirable that they have 67, considering the balance with  $\alpha$ -phase and the solidification temperature.

[0086] Meanwhile, in order to obtain mechanical properties such as ductility, impact strength, dezincification corrosion resistance, stress corrosion cracking resistance and machinability, the values of Expressions (1) to (4) are required to be 60 or more, and it is preferable that the values be 62.5 or more, more preferable that the values be 64 or more. On the other hand, in order to obtain high strength and wear resistance as well as excellent machinability, the values of Expressions (1) to (4) are required to be 71 or less, and it is preferable that the values be 68.5 or less. Furthermore, in order to obtain an industrially satisfactory machinability without Pb contained, it is most preferable that the values be 67 or less.

[0087] The copper-based alloy casting according to the invention with a phase structure, more than 80% of which is occupied by  $\alpha$ ,  $\kappa$  and  $\gamma$ -phases, can be obtained by the following casting conditions.

[0088] First, the maximum pouring temperature is, in general, 1150°C or less or liquidus temperature + 250°C or less, preferably 1100°C or less, and most preferably 1050°C



or less. The minimum pouring temperature is not specified as long as the molten alloy can reach every corner of the mold. However, in general, the minimum pouring temperature is in the range of 900 to 950°C, right above the liquidus temperature. It should be understood that the above temperature varies with the compositions of the alloys.

[0089] As described above, the phase structure has a close relationship with the above expressions, and the temperature range from solidification-complete temperature to 500°C has the biggest effect on the phase transformation. When the above expressions have the values of 62.5 or less, it is difficult to obtain a phase structure, more than 80% of which is occupied by  $\alpha$ ,  $\kappa$  and  $\gamma$ -phases, if the alloy is cooled at the average rate of 250°C/second or more. It is preferable to cool the alloy at the rate of 100°C/second or less when the expressions have the values of 62.5 or less. Meanwhile, if the above expressions have the values of 68.5 or more, and the alloy is cooled at the average rate of 0.5°C/second or less in the temperature range of 700 to 800°C, even though  $\alpha$ ,  $\kappa$  and  $\gamma$ -phases occupy more than 80% of the phase structure, the precipitations of  $\kappa$  and  $\gamma$ -phases are hindered, and thus  $\alpha$ -phase grains are grown, whereby it become more difficult to achieve the grain refinement. Therefore, it is preferable to cool the alloy at the rate of 1°C/minute, at least, in the temperature range of 700 to



800°C even when the expressions have the values of 68.5 or more.

[0090] It is needless to say that, in the alloys of the invention, the grains can be refined by common methods or means for casting refinement, that is, the decrease in pouring temperature, fastening the cooling rate, stirring during solidification or the like.

[0091] In the specification, the word 'casting' means a substance, which is wholly or partially melted and solidified. The casting includes various substances, beginning with ingot, slab, billet for rolling or extrusion, for example, castings by virtue of sand casting, metal casting, low-pressure casting, diecast, lost wax, semi-solid casting (for example, Thixo casting, Rheocasting), squeeze, centrifugal casting, continuous casting (for example, horizontal continuous casting, metallizing, build-up spraying or upward, rod manufactured by upCast, hollow rod, heteromorphic rod, heteromorphic hollow rod, coil material, wire material or the like), melting and forging (direct forging), metallizing, build-up spraying, lining, overlay. Also, in a broad sense, welding should be included in the casting because part of the base material is melted, solidified and combined together in welding.

[Embodiment]

[0092] Alloy materials having compositions shown in Tables



1 to 3 are melted in an electric furnace and poured into a metal mold in order to obtain specimens. The pouring temperature is 1000°C, and the pre-heating temperature of the metal mold is 200°C. The specimens are cylindrical, 40 mm in diameter and 280 mm in length.

[0093] The area ratios of respective phases composing the phase structure of the specimens are measured. Also, the cylindrical specimens are cut parallel to the bottom surface at 100 mm away from the bottom surface, and the mean grain sizes are measured at 10 mm away from the center of the cross-sectional surface of the specimen. The mean grain sizes are measured on the basis of the comparative methods for estimating average grain size of wrought copper and copper alloy of JIS H0501, in which, after the cut surfaces are etched by nitric acid, grains as large as 0.5 mm and more are observed with the naked eye or a magnifying glass offering 5 times the magnification, and grains smaller than 0.5 mm are etched by a mixed solution of hydrogen peroxide and ammonia water and then observed with an optical microscope. Meanwhile, the grain sizes are measured at 10 mm away from the axis of the cut surface and 100 mm away from the bottom surface.

[0094] Tables 1 to 3 disclose the measured grain sizes. Specimens No. 1 to 44 disclosed in Tables 1 and 2 are embodiments of the invention, and Specimens No. 101 to 122



disclosed in Table 3 are comparative examples. Among the comparative examples, bold-lettered data illustrate that the specimens do not follow the conditions defined for the copper-based alloy casting of the invention.



[Table 1]

No.	Chemical composition of alloys (remainder Zn and inevitable impurities) (mass %)										P/Zr	Si/Zr	Si/P	Expression		Area ratio of phase structure (%)		Mean grain size ( $\mu\text{m}$ )
	Cu	Si	Zr	P	Mg,B,C,Ti,REM	Al,Mn,Cr	Sn,Sb,As	Pb,Bi,Se,Te										
1	71.0	2.93	0.0150	0.10	-	-	-	-	-	6.7	195	29	(1)	60.4	85	15	100	
2	74.2	3.73	0.0160	0.10	-	-	-	-	-	6.3	495	78	(1)	60.8	85	15	90	
3	70.3	2.50	0.0120	0.12	-	-	-	-	-	10.0	208	21	(1)	61.2	90	10	80	
4	72.0	2.54	0.0155	0.07	-	-	-	-	-	4.5	164	36	(1)	62.9	95	5	65	
5	74.7	3.50	0.0180	0.09	-	-	-	-	-	5.0	194	39	(1)	63.9	100	0	30	
6	75.3	2.98	0.0007	0.09	-	-	-	-	-	129	4257	33	(1)	64.6	100	0	85	
7	75.8	3.10	0.0190	0.08	-	-	-	-	-	4.2	163	39	(1)	64.7	100	0	30	
8	75.9	3.08	0.0053	0.06	-	-	-	-	-	11.3	581	51	(1)	64.9	100	0	25	
9	75.8	3.00	0.0100	0.10	-	-	-	-	-	10	300	30	(1)	65.0	100	0	15	



10	76.1	3.10	0.0290	0.07	-	-	-	-	-	2.4	107	44	(1)	65.0	100	0	35
11	76.2	3.10	0.0017	0.07	-	-	-	-	-	58	2583	44	(1)	65.1	100	0	50
12	76.3	3.09	0.0185	0.07	-	-	-	-	-	3.8	167	44	(1)	65.3	100	0	25
13	76.1	3.00	0.0038	0.13	-	-	-	-	-	3.4	79	23	(1)	65.2	100	0	80
14	76.6	3.07	0.0040	0.08	-	-	-	-	-	20	768	38	(1)	65.6	100	0	20
15	81.0	3.80	0.0170	0.06	-	-	-	-	-	3.5	224	63	(1)	67.5	100	0	50
16	75.8	2.27	0.0280	0.08	-	-	-	-	-	2.9	81	28	(1)	67.6	100	0	65
17	83.1	4.21	0.0230	0.03	-	-	-	-	-	1.3	183	140	(1)	68.3	100	0	70
18	79.2	2.76	0.0210	0.16	-	-	-	-	-	7.6	131	17	(1)	69.1	100	0	75
19	80.2	2.70	0.0230	0.07	-	-	-	-	-	3.0	117	39	(1)	70.5	100	0	80
20	79.4	2.30	0.0160	0.11	-	-	-	-	-	6.9	144	21	(1)	71.0	100	0	90
21	76.9	3.20	0.0009	0.	Mg:0.004	-	-	-	-	88.9	3556	40	(2)	65.5	100	0	40
				08													
22	75.8	2.98	0.0032	0.07	Mg:0.11	-	-	-	-	21.9	931	43	(2)	65.2	100	0	20



23	73.8	2.76	0.0075	0.12	B:0.011	-	-	-	-	16.0	368	23	(2)	63.8	100	0	20
24	77.3	3.41	0.0110	0.09	C:0.001	-	-	-	-	8.2	310	38	(2)	65.1	100	0	15
25	75.9	3.00	0.0130	0.11	Ti:0.012	-	-	-	-	8.5	231	27	(2)	65.1	100	0	15

Reference: \* Expressions (1)  $Cu - 3.5 \times Si - 3 \times P$

(2)  $Cu - 3.5 \times Si - 3 \times P - 0.5 \times [i] + 0.5 \times [ii]$

(3)  $Cu - 3.5 \times Si - 3 \times P - 1.8 \times Al + a \times Mn + 0.5 \times Cr$

(4)  $Cu - 3.5 \times Si - 3 \times P - 0.5 \times [i] + 0.5 \times [ii] - 1.8 \times Al + a \times Mn + 0.5 \times Cr$



[Table 2]

No.	Chemical composition of alloys (remainder Zn and inevitable impurities) (mass %)										P/Zr	Si/Zr	Si/P	Expression		Area ratio of phase structure (%)		Mean grain size ( $\mu\text{m}$ )
	Cu	Si	Zr	P	Mg, B, C, Ti, REM	Al, Mn, Cr	Sn, Sb, As	Pb, Bi, Se, Te						type	value	$\alpha+\kappa+\gamma$	others	
26	76.6	3.12	0.0150	0.08	REM: 0.05						5.3	208	39	(2)	65.4	100	0	20
27	75.2	3.12	0.0035	0.09	-	Mn: 0.4	-	-			26	891	35	(3)	64.2	100	0	30
28	70.9	4.53	0.0085	0.17	-	Mn: 3.6	-	-			20	533	27	(3)	61.7	95	5	40
29	73.3	4.02	0.0120	0.15	-	Al: 0.5 Mn: 2.7	-	-			13	335	27	(3)	63.3	100	0	25
30	75.3	3.65	0.0160	0.10	-	Al: 0.9 Mn: 0.9					6	228	37	(3)	62.4	100	0	35
31	75.6	3.13	0.0240	0.10	C: 0.0006	Cr: 0.2	-	-			4	130	31	(4)	64.4	100	0	30
32	74.9	2.89	0.0035	0.11	-	-	Sn: 0.15	-			31	826	26	(1)	64.5	100	0	25



33	78.4	3.12	0.0140	0.08	-	-	-	Sn: 1.4	-	6	223	39	(1)	67.2	100	0	15
34	78.8	3.76	0.0035	0.13	-	-	-	Sb: 0.03	-	37	1074	29	(1)	65.2	100	0	30
35	76.5	3.11	0.0015	0.03	-	-	-	As: 0.13	-	20	2073	104	(1)	65.5	100	0	50
36	76.8	3.12	0.0230	0.08	-	-	-	-	Pb: 0.08	3	136	39	(1)	65.7	100	0	30
37	76.2	3.08	0.0125	0.07	-	-	-	-	Bi: 0.06	6	246	44	(1)	65.2	100	0	25
38	75.6	2.99	0.0180	0.05	-	-	-	-	Bi: 0.3	3	166	60	(2)	65.0	100	0	25
									Se: 0.3								
39	76.7	3.06	0.0180	0.12	-	-	-	Sn: 0.6	Pb: 0.015	6	170	28	(1)	65.7	100	0	20
40	82.3	3.80	0.0150	0.04	-	Al: 1.2	-	-	Bi: 0.25	3	253	95	(3)	66.7	100	0	25
41	73.2	3.82	0.0095	0.12	Mg: 0.008	Mn: 1.9	-	-	Pb: 0.19				(4)	66.3	100	0	20
42	74.5	3.98	0.0055	0.09	Mg: 0.032	Al: 0.04	Sn: 0.8	-	-	15	727	44	(4)	66.0	100	0	15
						Mn: 2.9											
43	78.8	3.22	0.0110	0.08	-	Al: 1.2	Sb: 0.09	-	-	7	293	40	(3)	65.1	100	0	15
44	74.7	3.50	0.0180	0.09	-	Al: 0.2	-	Pb: 0.15	-	5	194	39	(3)	64.0	100	0	30







[Table 3]

No.	Chemical composition of alloys (remainder Zn and inevitable impurities) (mass %)										P/Zr	Si/Zr	Si/P	Expression		Area ratio of phase structure (%)		Mean grain size (μm)
	Cu	Si	Zr	P	Mg, B, C, Ti, REM	Al, Mn, Cr	Sn, Sb, As	Pb, Bi, Se, Te	Fe, Ni									
101	70.2	4.45	0.0100	0.08	-	-	-	-	-	-	8.0	445	56	(1)	54.4	60	40	1500
102	73.0	3.98	0.0150	0.10	-	-	-	-	-	-	6.7	265	40	(1)	58.8	65	35	800
103	70.3	3.08	0.0310	0.10	-	-	-	-	-	-	3.2	99	31	(1)	59.2	60	40	800
104	69.3	2.64	0.0170	0.11	-	-	-	-	-	-	6.5	155	24	(1)	59.7	70	30	600
105	79.5	2.10	0.0030	0.07	-	-	-	-	-	-	23.3	700	30	(1)	71.9	95	5	300
106	86.0	4.12	0.0290	0.09	-	-	-	-	-	-	3.1	142	46	(1)	71.3	100	0	200
107	82.5	2.56	0.0120	0.08	-	-	-	-	-	-	6.7	213	32	(1)	73.3	100	0	250
108	82.7	2.25	0.0055	0.10	-	-	-	-	-	-	4.2	93	22	(1)	74.5	100	0	300



109	79.8	4.05	-	-	-	-	-	-	-	-	-	-	-	-	(1)	65.6	100	0	2000
110	76.2	3.12	0.0003	0.09	-	-	-	-	-	-	300	10400	35	(1)	65.0	100	0	500	
111	76.1	3.07	0.0002	0.07	-	-	-	-	-	-	350	15350	44	(1)	65.1	100	0	600	
112	74.7	2.95	0.0500	0.09	-	-	-	-	-	-	1.8	59	33	(1)	64.1	100	0	150	
113	72.8	2.35	0.1500	0.08	-	-	-	-	-	-	0.5	16	29	(1)	65.5	100	0	200	
114	79.3	4.05	0.3000	0.03	-	-	-	-	-	-	0.1	14	135	(1)	66.4	100	0	200	
115	75.6	3.18	0.0050	0.005	-	-	-	-	-	-	1.0	636	64	(1)	64.5	100	0	350	
116	70.2	1.70	0.0060	0.08	-	-	-	-	-	-	13.3	283	21	(1)	64.0	95	5	200	
117	85.8	5.50	0.0110	0.10	-	-	-	-	-	-	9.1	500	55	(1)	66.3	100	0	200	
118	76.6	3.11	0.0180	0.09	-	-	-	-	-	-	Fe: 5.0	173	35	(1)	65.4	100	0	400	
										0.55									
119	75.8	3.05	0.0170	0.09	-	-	-	-	-	-	Ni: 5.3	179	34	(1)	64.9	100	0	600	
											0.6								
120	70.1	2.77	0.0180	0.08	-	-	-	-	-	-	4.4	154	35	(1)	60.2	75	25	500	



121	72.9	3.45	0.0150	0.15	-	-	-	-	-	10.0	230	23	(1)	60.4	75	25	400
122	76.5	3.05	-	0.08	-	-	Sn: 0.6	Pb: 0.015	-	-	-	-	(2)	65.6	100	0	1500

Reference: \* Expressions (1)  $Cu - 3.5 \times Si - 3 \times P$

(2)  $Cu - 3.5 \times Si - 3 \times P - 0.5 \times [i] + 0.5 \times [ii]$

(3)  $Cu - 3.5 \times Si - 3 \times P - 1.8 \times Al + a \times Mn + 0.5 \times Cr$

(4)  $Cu - 3.5 \times Si - 3 \times P - 0.5 \times [i] + 0.5 \times [ii] - 1.8 \times Al + a \times Mn + 0.5 \times Cr$



[0095] First, the phase structure will be described.

[0096] Specimens No. 1 to 3 of the embodiments and Specimens No. 120 and 121 of the comparative examples have almost the same values for the expressions, and the results disclose that the mean grain sizes decrease as the total area ratios of  $\alpha$ ,  $\kappa$  and  $\gamma$ -phases increase. It can be found out that the above three phases should occupy more than 80% of the phase structure in order to refine the grains as small as 100  $\mu\text{m}$  or less, which is an object of the invention.

[0097] In addition, the phase structures of Specimen No. 9 of the embodiments and Specimen No. 103 of the comparative examples are disclosed in Figs. 1 and 2. In Fig. 1,  $\alpha$ ,  $\kappa$  and  $\gamma$ -phases occupy 100% of the phase structure, and the mean grain size is 15  $\mu\text{m}$ . In Fig. 2,  $\alpha$ ,  $\kappa$  and  $\gamma$ -phases occupy 60% of the phase structure with  $\beta$ -phase occupying the remainder, and the mean grain size is 800  $\mu\text{m}$ .

[0098] It is evident from Tables 1 and 2 that the mean grain size is refined as small as 100  $\mu\text{m}$  or less after melt-solidification if the amount of Zr, the values of the expressions, the area ratios of  $\alpha$ ,  $\kappa$  and  $\gamma$ -phases satisfy the conditions defined for the invention.

[0099] In Specimens No. 101 to 104 of the comparative examples, the expressions have values of less than 60, the total area ratios of the above three phases are less than 80%, and the mean grain sizes are considerably large.



[0100] In Specimens No. 105 to 108 of the comparative examples, the expressions have values of larger than 71. Even though the other conditions defined for the invention are satisfied, the mean grain sizes are 200  $\mu\text{m}$  or larger.

[0101] In Specimens No. 120 and 121 of the comparative examples, even though the total area ratios of the three phases are less than 80% and the values of the expressions approaches the minimum defined for the invention, the mean grain sizes are 400  $\mu\text{m}$  or larger.

[0102] Specimen No. 109 of the comparative examples contains no Zr and P, and Specimens No. 110 and 111 of the comparative examples contain a smaller amount of Zr than that defined for the invention. Meanwhile, since Specimens No. 110 and 111 of the comparative examples contain a small amount of Zr, the values of Si/Zr and P/Zr are not in the preferable ranges of the invention, and thus the mean grain sizes are considerably large.

[0103] Specimens No. 113 to 115 of the comparative examples contain larger amounts of Zr than that defined for the invention, and it can be found out that the grain refinement is hindered if more than 0.05% of Zr is contained.

[0104] Figs. 9A and 9B show the relationship of the mean grain size with the amount of Zr for the specimens of the embodiments (Cu, Si, Zr, P and the remainder Zn), for which the expressions have the values in the preferable range of



64 to 67, and Specimens No. 110 to 115 of the comparative examples. In Figs. 9A and 9B, the values of the expressions are limited within the range shown in the figures because, as shown in Specimens No. 1 to 4 and No. 15 to 20 of the embodiment, the values of the expressions considerably influence the mean grain sizes outside the limited range. Therefore, in the limited range, the mean grain sizes are estimated without influence by the values.

[0105] Specimen No. 115 of the comparative examples contains a smaller amount of P than that defined for the invention. Also, in Specimens No. 116 and 117 of the comparative examples, the amounts of Si are not in the defined range of the invention, and the mean grain sizes are 200  $\mu\text{m}$  or more.

[0106] Specimens No. 118 and 119 of the comparative examples illustrate that the mean grain size increase when Fe and Ni are contained more than defined for the invention as impurities.

[0107] Figs. 10 to 12 illustrate respectively the metal structures of Specimen No. 8 of the embodiments (25  $\mu\text{m}$  in the mean grain size), Specimens No. 115 (350  $\mu\text{m}$  in the mean grain size) and 110 (500  $\mu\text{m}$  in the mean grain size) of the comparative examples that are cooled during solidification process, specifically at which 40% solid-phases and 60% liquid-phases coexist (semi-molten state), and then etched.



[0108] During solidification (melt-solidification), in Specimen No. 8 of the embodiments, the arms of dendrites are not generated, and thus the dendrites have a circular or oval shape, contrary to the above, in Specimens No. 115 and 110 of the embodiments, dendrites have a tree-shape. Like the above, in Specimen No. 8 of the embodiments, crystal nuclei are generated faster than grain growth (growth of the arms of dendrites), whereby the grains can be refined (the matrix is a liquid phase in the semi-molten state).

[0109] The above fact illustrates that the copper-based alloy casting of the invention is preferable for semi-solid casting, and if the solid phase is granular, both liquid and solid phases can reach every corner of the metal mold without substantial resistance.

[0110] In order to evaluate the castability of the copper-based alloy castings of the invention, in which the grains are refined, Tatur Shrinkage Test is performed on the specimens illustrated in Table 4, and the shapes of the inside shrinkages and the existence of defects such as porosity, hole, cavity or the like in the vicinity of the inside shrinkages is examined. The castability is evaluated 'good' for specimens with smooth shapes of the inside shrinkages and no defects such as porosity or the like at the final solidification portion as shown in Fig. 13A, 'fair' for specimens with non-smooth shapes of the inside



shrinkages and some defects such as porosity or the like at the final solidification portion as shown in Fig. 13B, and 'bad' for specimens with uneven shapes of the inside shrinkage and obvious defects such as porosity or the like at the final solidification portion as shown in Fig. 13C. Table 4 illustrates the test result.

[Table 4]

Specimen No.	Mean grain size	Tatur shrinkage test
6	85 $\mu\text{m}$	Good
9	15 $\mu\text{m}$	Good
102	800 $\mu\text{m}$	Bad
108	300 $\mu\text{m}$	Bad
109	2000 $\mu\text{m}$	Bad
110	500 $\mu\text{m}$	Fair
113	200 $\mu\text{m}$	Fair

[0111] As illustrated in Table 4, Specimens No. 6 and 9 of the embodiments show excellent castability, however, the castibilities becomes fair or bad when the mean grain sizes become 200  $\mu\text{m}$  or more.

[0112] Meanwhile, Figs. 14 and 15 illustrate the results of Specimens No. 9 and 109. It is evident from the comparison between Figs. 14B and 14C and Figs. 15B and 15C



that casting defects cannot be discovered in Specimen No. 9 in which the grains are refined, on the contrary, in Specimen No. 109, cracks, cavities and lots of holes are discovered among the arms of dendrites, shrinkage cavities are large, unevenness at the final solidification portion is severe, and casting defects are included.

[0113] Next, in order to evaluate the characteristics of the copper-based alloy castings of the invention in which the grains are refined, mechanical properties (tensile strength, proof stress, elongation and fatigue strength) are measured for Specimens No. 8, 9, 12, 29, 39, 44, 122, 110, 111 and 112.

[0114] In Specimens No. 110, 111 and 112, the specimens are further heated up to 750°C and hot-extruded on the condition of the extrusion ratio of 9 and the rolling reduction of 89% so as to fabricate round bars with diameters of 13.3 mm, and then the mean grain sizes and mechanical properties are measured. Meanwhile, hot-extruded specimens are indicated with Specimens No. 110a, 111a, and 112a.

[0115] As a mechanical property test, Specimen No. 10 regulated by JIS Z 2201 is adopted among specimens, and tensile test is performed with an Amsler universal testing machine in order to measure the tensile strength, proof stress (0.2%), elongation and fatigue strength. Table 5



illustrates the result.

[Table 5]

Specimen No.	Mean grain size ( $\mu\text{m}$ )	Tensile strength ( $\text{N/mm}^2$ )	0.2% proof stress ( $\text{N/mm}^2$ )	Elongation (%)	Fatigue strength ( $\text{N/mm}^2$ )
8	25 (after melt- solidification)	516	257	42	255
9	15 (after melt- solidification)	526	274	42	261
12	25 (after melt- solidification)	520	263	40	257
29	25 (after melt- solidification)	652	345	24	330
39	20 (after melt- solidification)	525	271	30	252
44	30 (after melt- solidification)	605	310	26	285
122	1500 (after melt- solidification)	388	184	15	159
110	500 (after melt- solidification)	436	181	26	169



110a	30 (after hot working)	500	254	37	250
111	600 (after melt-solidification)	433	174	24	155
111a	30 (after hot working)	498	251	36	248
112	150 (after melt-solidification)	452	199	30	186
112a	20 (after hot working)	524	272	36	258

[0116] Referring to Table 5, Specimens No. 8, 9, 12, 29, 39 and 44 of the embodiments have better mechanical properties than Specimens No. 122, 110, 111 and 112 of the comparative examples. In addition, since containing Al and Mn, Specimens No. 22 and 44 have better mechanical properties than the other specimens of the embodiments.

[0117] It can be found out from the comparison between Specimen No. 39 containing 0.6% of Sn, a low-melting point metal, and Specimen No. 122 that mechanical properties such as strength and elongation, in particular, are considerably improved by the grain refinement, which is attributed to the



addition of Zr and P.

[0118] In Specimens No. 110 to 112, the grains are large after melt-solidification, however, the grains can be refined as small as 30  $\mu\text{m}$  or less by hot extrusion. Also, mechanical properties of the specimens in which the grains are refined by hot extrusion are almost equal or still inferior to those of the melt-solidified embodiments. It is evident from the above results that the mechanical properties are dependent on the mean grain size. Therefore, it can be found out that the copper-based alloy castings according to the invention, in which the grains are being refined during the melt-solidification, can have the mechanical properties as excellent as those of specimens that are hot-extruded, even though they are not hot-extruded.

[0119] The corrosion resistance (erosion · corrosion test, dezincification corrosion test and stress corrosion cracking test) of the specimens shown in Table 6 is examined.

[0120] With the erosion · corrosion test, samples taken from the specimens are continuously in contact with a 3% saline solution of 30°C at a rate of 11 m/s by a 2 mm-diameter nozzle, and then the weight loss due to corrosion is measured after 48 hours. Table 7 illustrates the results.

[0121] The dezincification test is based on ISO 6509. Samples taken from the specimens are buried in phenol-resin materials, polished by emery papers up to No. 1200, washed



by ultrasonic waves in pure water, and then dried. Samples for the corrosion resistance test obtained by the above procedure are soaked in an aqueous solution (12.7 g/l) of 1.0% cupric chloride dihydrate ( $\text{CuCl}_2 \cdot 2\text{H}_2\text{O}$ ), maintained for 24 hours at 75°C, taken out from the aqueous solution, and then the maximum values of the dezincification corrosion depth (maximum dezincification corrosion depth) are measured. Table 7 illustrates the results.

[0122] The stress corrosion cracking test is based on JIS H3250. Tabular samples (10 mm in width, 60 mm in length and 5 mm in depth) taken from the specimens are bent (to add residual tensile stress) at 45 degree so as to have V-shapes (the radius of the bent area is 5 mm), fat-removed, dried, and maintained under the ammonia atmosphere (25°C) in a desiccator containing 12.5% ammonia water (ammonia diluted with the same amount of pure water). The samples are taken out from the desiccator after predetermined time described below, washed with 10% sulfuric acid, and then the existence of cracks in the samples are observed by a magnifying glass (10 times magnification). Table 6 illustrates the observation results. In Table 6, specimens, in which cracks are not found after 2 hour-maintenance in the ammonia atmosphere, however, found after 8 hour-maintenance, are expressed with 'x', specimens, in which cracks are not found after 8 hour-maintenance, however, found after 24 hour-



maintenance, are expressed with ' $\Delta$ ', and specimens, in which cracks are not found after 24 hour-maintenance, are expressed with 'o'.

[Table 6]

Specimen No.	Mean grain size ( $\mu\text{m}$ )	Weight reduction due to corrosion ( $\text{mg}/\text{cm}^2$ )	Maximum corrosion depth ( $\mu\text{m}$ )	Stress corrosion cracking resistance
8	25	14.8	Less than 10	o
9	15	15.2	Less than 10	o
15	50	14.1	Less than 10	o
42	15	7.9	Less than 10	o
33	15	5.6	Less than 10	o
103	800	29.2	280	x
115	350	18.5	180	$\Delta$

[0123] Referring to Table 6, Specimens No. 8, 9, 15, 42 and 33 of the embodiments in which the grains are refined,



have better corrosion resistance than Specimens No. 103 and 115 of the comparative examples. In addition, since containing corrosion resistance-improving elements, Specimens No. 42 and 33 are superior in the weight loss due to corrosion, in particular, to the other specimens of the embodiments.

[0124] The outer circumferential surfaces of the specimens shown in Table 7 are cut by a lathe provided with a point nose straight tool (rake angle:  $-6^{\circ}$ , nose R: 0.4 mm) on the condition of the cutting speed of 100 m/minute, the cutting depth of 1.5 mm, the feed of 0.11 mm/rev, power is measured with a three component dynamometer attached to the bite, and calculated into the primary cutting force. Also, the machinability is evaluated from the shapes of chips generated during cutting. That is, when the chips are shaped like fan or circular-arc, that is, half-rotating, the treatability is good, and the specimens are expressed with  $\odot$ . When the chips are shaped like a fine needle, the treatability is good, but there are some dangers in that the debris forms obstacles to machine tools such as lathe or the like, and the debris may be stuck into the operator's finger, whereby the specimens are expressed with 'o'. On the other hand, when the debris are shaped like a screw, that is, rotating over three times, there are disadvantages in that cutting treatability deteriorates, debris may be stuck into



the bite, and the cutting surface can be damaged, whereby the specimens are expressed with 'x'. Meanwhile, when the debris are shaped like a circular-arc rotating more than half, but, less than one time, or a screw rotating less than three times, even though considerable troubles do not occur, the treatability of the chips deteriorates, the debris may be stuck into the bite during continuous cutting, and then the cutting surfaces are damaged, whereby the specimens are expressed with ' $\Delta$ '.

[0125] In addition, with the surface roughness, it is ideal that  $R_y$  approaches the theoretical surface roughness, and the specimens are expressed with 'o' when  $R_y$  are less than 7.5  $\mu\text{m}$ . Also, in order to obtain the industrially satisfying cutting surfaces, the specimens are expressed with ' $\Delta$ ' when  $R_y$  are in the range of 7.5 to 12  $\mu\text{m}$ , and the specimens are expressed with 'x' when  $R_y$  are more than 12  $\mu\text{m}$ .

[Table 7]

Specimen No.	Mean grain size	Main cutting force	Shape of chips	Surface state
8	25 $\mu\text{m}$	118 N	$\odot$	o
36	30 $\mu\text{m}$	112 N	$\odot$	o
39	20 $\mu\text{m}$	114 N	$\odot$	o
103	800 $\mu\text{m}$	161 N	x	$\Delta$



107	250 $\mu\text{m}$	185 N	$\times$	$\Delta$
110	500 $\mu\text{m}$	121 N	$\odot$	$\Delta$
113	200 $\mu\text{m}$	135 N	$\Delta$	$\times$

[0126] Referring to Table 7, Specimens No. 8, 36 and 39 of the embodiments in which the grains are refined, have better machinability than Specimens No. 103, 107, 110 and 113 of the comparative examples. Meanwhile, since containing machinability-improving elements, Specimens No. 36 and 39 have smaller primary cutting forces than Specimen No. 8.

[0127] The copper-based alloy casting according to the invention, in which the grains are refined during the melt-solidification, can be used as the following structural materials:

- Common mechanical parts requiring castability, conductivity, thermal conductivity and high mechanical properties;
- Electric terminal and connector requiring high degree of conductivity and thermal conductivity, electric parts, on which alloy brazing and welding can be easily performed;
- Gauge parts requiring easy castability;
- Metal fittings for water supply and drainage, metal fittings for construction, daily necessities • miscellaneous goods requiring excellent mechanical properties;



- Marine propeller, shaft, bearing, valve sheet, valve rod, metal fitting for wrench, cramp, metal fitting for connection, door knob, pipe buckle and cam requiring high strength, hardness and excellent corrosion resistance, toughness;

- Valve, stem, bush, worm gear, arm, cylinder parts, valve sheet, stainless bearing and pump impeller requiring high degree of strength, hardness and wear resistance;

- Valve, pump chassis, paddle wheel, hydrant, combination tap, water line valve, joint, spring cooler, cock, water meter, water stop valve, sensor parts, scroll type compressor parts, high pressure valve and sleeve pressure vessel requiring pressure resistance, wear resistance, machinability, and castability;

- Sliding parts, hydraulic cylinder, cylinder, gear, fishing reel and fastener of airplane requiring excellent hardness and wear resistance

- Bolt, nut and pipeline connector requiring excellent strength, corrosion resistance and wear resistance;

- Chemical machine parts and industrial valve suitable for large sized casting with simple shape and requiring excellent strength, corrosion resistance and wear resistance;

- Welded pipe of fresh water generator or the like, hydrant, heat exchanger pipe, heat exchanger tube sheet, gas



pipeline pipe, elbow, marine structural material, welding member and welding material requiring bond strength, build-up spraying, lining, overlay, corrosion resistance and castability;

- A valve, hexagon cap nut and header hydrant parts requiring excellent stress resistance, wear resistance and machinability;

- Sliding bearing requiring malleability, ductility, fatigue resistance and corrosion resistance, heat exchanger, heat exchanger tube sheet and marine parts requiring corrosion resistance and sea water resistance;

- Spindle or structural material requiring excellent machinability, malleability and ductility.

[0128] Copper-based alloy castings according to the invention have the above compositions and phase structures, and grains are refined to be about 100  $\mu\text{m}$  or less in the mean grain size after melt-solidification.

[0129] Since the grains are being refined during the melt-solidification, the castings can endure shrinkage during solidification, and casting cracks hardly occur. In addition, since holes and porosities, generated during solidification, can be removed easily, robust castings without casting defects such as cavities, shrinkage cavities or the like can be fabricated.

[0130] In addition, dendrites, crystallized during



solidification, have no arms, different from the typical dendrite structure in a casting, that is, a tree-like shape, and are preferably shaped circular, oval, and polygonal or like a cruciform. Therefore, the fluidity of molten alloy improves, and the molten alloy can reach every corner of a thin-walled and complex-shaped mold.

[0131] Since having excellent proof stress and the other mechanical properties, corrosion resistance, machinability or the like, the castings, grains of which are refined, are particularly useful for As-Cast products with complex shapes such as valve, joint, water faucet, metal fitting for water supply and drainage or the like.